

# Synthesis of nanocrystals in $\text{KNb}(\text{Ge},\text{Si})\text{O}_5$ glasses and chemical etching of nanocrystallized glass fibers

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## Abstract

The nanocrystallization behavior of  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-(50-x)\text{GeO}_2-x\text{SiO}_2$  glasses with  $x = 0, 25,$  and  $50$  (i.e.,  $\text{KNb}(\text{Ge},\text{Si})\text{O}_5$  glasses) and the chemical etching behavior of transparent nanocrystallized glass fibers have been examined. All glasses show nanocrystallization, and the degree of transparency of the glasses studied depends on the heat treatment temperature. Transparent nanocrystallized glasses can be obtained if the glasses are heat treated at the first crystallization peak temperature. Transparent nanocrystallized glass fibers with a diameter of about  $100\ \mu\text{m}$  in  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-50\text{GeO}_2$  are fabricated, and fibers with sharpened tips (e.g., the taper length is about  $450\ \mu\text{m}$  and the tip angle is about  $12^\circ$ ) are obtained using a meniscus chemical etching method, in which etching solutions of 10 wt% -HF/hexane and 10 M -NaOH/hexane are used. Although the tip (aperture size) has not a nanoscaled size, the present study suggests that  $\text{KNb}(\text{Ge},\text{Si})\text{O}_5$  nanocrystallized glass fibers have a potential for new near-field optical fiber probes with high refractive indices of around  $n = 1.8$  and high dielectric constants of around  $\epsilon = 58$  (1 kHz, room temperature).

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**Keywords:** Nanocrystallization; Raman scattering spectra; Glass fiber; Chemical etching; Fiber probes

## 1. Introduction

Nanostructures are the gateway into a realm in physical, chemical, biological and materials science. Developing techniques for fabricating nanostructures inexpensively is an area that requires substantial effort. Crystallization of glass is one of the effective methods for fabrication of nanostructures [1]. Recently, transparent nanocrystallized glasses consisting of nonlinear optical/ferroelectric nanocrystals have received much attention, because such materials have a high potential for applications in photonic devices [2–6]. For instance,  $\text{TeO}_2$ -based nanocrystallized glasses showing a second harmonic generation (SHG) have been successfully fabricated [2–4].

Very recently, it has been found that a glass with the composition of  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-50\text{GeO}_2$  (i.e.,  $\text{KNbGeO}_5$ ) shows a prominent nanocrystallization [7–9]. Although second-order optical nonlinearities have not been detected from transparent nanocrystallized  $\text{KNbGeO}_5$  glasses,

elastic and mechanical properties of  $\text{KNbGeO}_5$  glass have largely improved through nanocrystallization [7,8]. The crystallization behavior of the glass system of  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{SiO}_2$  has been also studied, and the crystallized glasses showing SHGs have been fabricated [10–15]. In particular, the possibility of the formation of nanocrystals showing SHGs in  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{SiO}_2$  glasses has been proposed [12,13,15]. Considering the features in crystallization and optical properties of  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{GeO}_2$  and  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{SiO}_2$  glass systems, it is of interest to study the crystallization behavior of  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{GeO}_2-\text{SiO}_2$  glasses, and information might lead to the fabrication of transparent nanocrystallized glasses showing SHGs in this glass system.

On the other hand, as practical applications of transparent nanocrystallized glasses, the fabrication of nanocrystallized glass fibers is of interest [16–18]. Furthermore, among nanoscale science and technology, scanning near-field optical microscope (SNOM) absorbs great interests, in which usually optical glass fiber probes with sharpened tips have been used [19–21]. At this moment, however, it is unclear whether commercially available  $\text{SiO}_2$ -based glass

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fibers are best for near-field optical fiber probes or not. Considering the physical concept for the generation of near-field light at around the surface of particle, nanoparticles or sharpened tip fibers consisting of materials with high refractive indices (high dielectric constants) are favorable for near-field optical probes. In such materials, the generation of high intense near-field light would be expected through the generation of electric dipoles under light irradiation [22]. The refractive index ( $n$ ) of  $\text{SiO}_2$  glass is  $n = 1.45$ , as well-known. On the other hand, transparent nanocrystallized  $\text{KNbGeO}_5$  glasses have the value of around  $n = 1.8$  [7–9]. In this point of view, it is of particular interest to fabricate optical fiber probes consisting of nanocrystals with high refractive indices, i.e., nanocrystallized glass fiber probes. There has been no report on nanocrystallized glass fiber probes and also chemical etching of glass fibers consisting of nanocrystals so far.

In this study, we focus our attention on the nanocrystallization of  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-(50-x)\text{GeO}_2-x\text{SiO}_2$  glasses (i.e.,  $\text{KNb}(\text{Ge,Si})\text{O}_5$  glasses) and the chemical etching of transparent nanocrystallized glass fibers. We clarified the formation behavior of nanocrystals in  $\text{KNb}(\text{Ge,Si})$  glasses and optical properties of nanocrystallized glasses. Furthermore, we found a possibility for the fabrication of nanocrystallized glass fiber probes with sharpened tips by using a chemical etching technique.

## 2. Experimental

Glass compositions examined in the present study are shown in Table 1. Commercial powders of reagent grade  $\text{K}_2\text{CO}_3$ ,  $\text{Nb}_2\text{O}_5$ ,  $\text{GeO}_2$  and  $\text{SiO}_2$  were mixed and melted in a platinum crucible at 1200–1450 °C for 1 h in an electric furnace. The melts were poured onto an iron plate and pressed to a thickness of about 1 mm with another iron plate. The glass transition,  $T_g$  and crystallization peak,  $T_p$  temperatures were determined using differential thermal analyzes (DTA) at a heating rate of 10 K/min. Densities of glasses,  $d$ , were determined with the Archimedes method using distilled water as an immersion liquid. Refractive indices at a wavelength of 632.8 nm (He–Ne laser),  $n$ , were measured at room temperature with an ellipsometer (Mizojiri Optical, DVA-36 model).

Glasses were heat treated at various temperatures for 3 h and crystalline phases present in the heat-treated samples

were examined by X-ray diffraction (XRD) analyzes at room temperature using  $\text{CuK}\alpha$  radiation for powder (pulverized) samples. Optical transmission spectra were taken in the wavelength range of 190–2500 nm on a Shimadzu UV-3150 spectrometer. Micro-Raman scattering spectra at room temperature for the precursor glasses and crystallized samples were measured with a laser microscope (Tokyo Instruments Co., Nanofinder) operated at  $\text{Ar}^+$  (488 nm) laser. In our micro-Raman apparatus, the data below  $\sim 300\text{ cm}^{-1}$  cannot be measured due to the use of edge filter. SHGs of crystallized powders at room temperature were examined using the powder method proposed by Kurtz and Perry [23]. A fundamental wave of a Q-switched Nd:YAG laser operating at a wavelength of  $\lambda = 1064\text{ nm}$  was used as the incident light.

Glass fibers with a diameter of  $\sim 100\text{ }\mu\text{m}$  were prepared by pulling up the melt using a silica glass rod, where the pulling rate of the silica rod is  $\sim 1\text{ m/s}$ . In this method, the fibers were prepared by a hand-operated technique, but not using any fiber drawing machine. Then, the glass fibers were heat treated to obtain transparent nanocrystallized glass fibers. Glass and nanocrystallized fibers were etched using a meniscus chemical etching method [24], in which etching solutions such as 10 M-NaOH/hexane and 10 wt%-HF/hexane were used.

## 3. Results and discussion

### 3.1. Characterization of $\text{KNb}(\text{Ge,Si})\text{O}_5$ glasses

The glasses with the chemical compositions of  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-50\text{GeO}_2$ ,  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-25\text{GeO}_2-25\text{SiO}_2$  and  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-50\text{SiO}_2$  were prepared in this study, and they are designated here as KNG, KNGS, and KNS, respectively. The DTA patterns for bulk KNG, KNGS and KNS glasses are shown in Fig. 1. All these glasses show clear endothermic peaks due to the glass transition and exothermic peaks due to the crystallization. The values of  $T_g$  and  $T_p$  are summarized in Table 1. It is seen that the glass transition and crystallization peak temperatures increase when the amount of  $\text{SiO}_2$  increases. Furthermore, it is noted that KNG glass shows one sharp crystallization peak, but two peaks are observed in KNS glass. KNGS glass has one crystallization peak, but its shape is broad. These results indicate that KNG and KNS glasses show different crystallization behaviors.

Table 1  
Glass compositions, values of glass transition  $T_g$ , crystallization peak  $T_p$  temperatures, density  $d$ , and refractive index  $n$  in  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{GeO}_2-\text{SiO}_2$  glasses examined in the present study

Sample	Composition (mol%)				$T_g$ (°C) ( $\pm 3$ °C)	$T_p$ (°C) ( $\pm 3$ °C)	$d$ (g/cm <sup>3</sup> ) ( $\pm 0.01$ )	$n$ ( $\pm 0.01$ )
	$\text{K}_2\text{O}$	$\text{Nb}_2\text{O}_5$	$\text{GeO}_2$	$\text{SiO}_2$				
KNG	25	25	50	—	619	679	3.83	1.81
KNGS	25	25	25	25	655	720	3.60	1.77
KNS	25	25	—	50	670	761	3.34	1.75

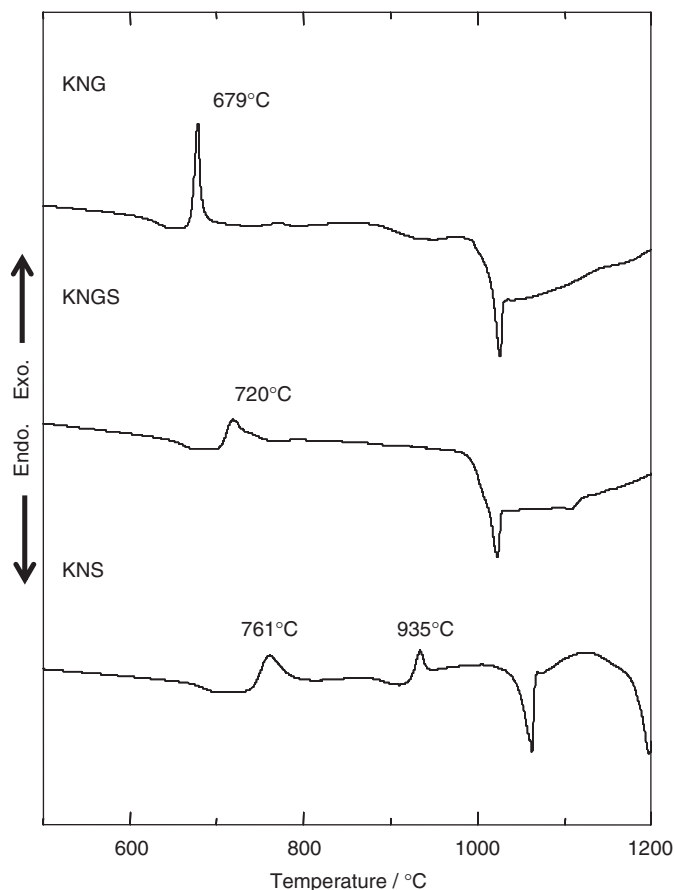


Fig. 1. DTA patterns for bulk  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-(50-x)\text{GeO}_2-x\text{SiO}_2$  glasses. Heating rate was 10 K/min. KNG for  $x = 0$ , KNGS for  $x = 25$ , and KNS for  $x = 50$ .

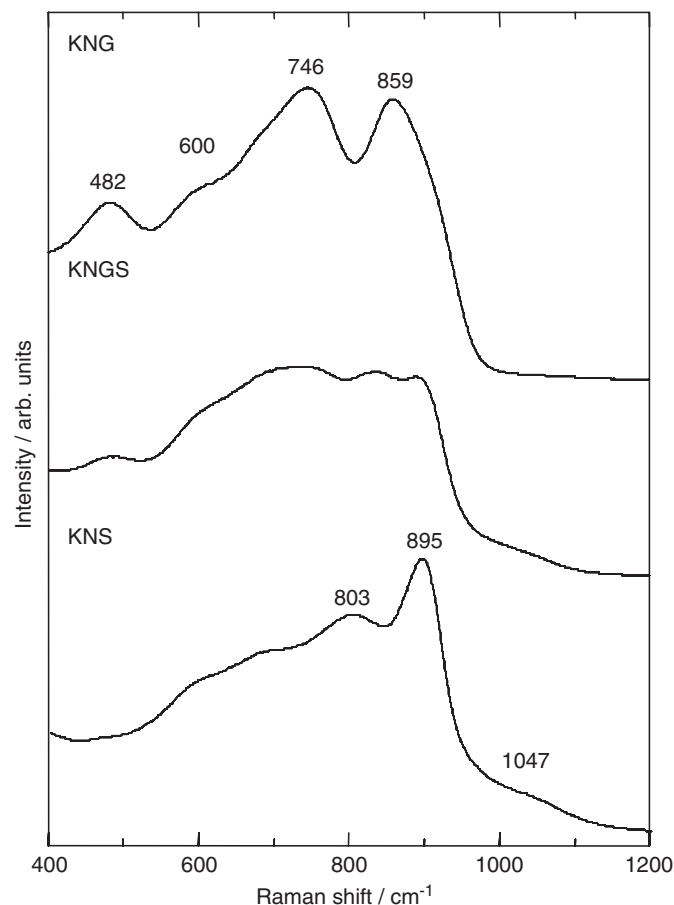


Fig. 2. Raman scattering spectra at room temperature for  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-(50-x)\text{GeO}_2-x\text{SiO}_2$  glasses. KNG for  $x = 0$ , KNGS for  $x = 25$ , and KNS for  $x = 50$ .

The values of  $d$  and  $n$  are also given in Table 1. It is clear that both values increase when the amount of  $\text{GeO}_2$  increases. Since the optical basicity ( $A = 0.70$  or  $0.63$ ), i.e., electron donor ability, of  $\text{GeO}_2$  oxide is larger than that ( $A = 0.48$ ) of  $\text{SiO}_2$  oxide [25–27], it is expected that KNG glass would give a large refractive index compared with KNS glass, as observed in this study. The optical basicity ( $A = 1.05$ ) of  $\text{Nb}_2\text{O}_5$  oxide is large [25], and thus all KNG, KNGS and KNS glasses with 25 mol%  $\text{Nb}_2\text{O}_5$  indicate relatively large refractive indices of  $n = 1.75$ – $1.81$ .

The Raman scattering spectra at room temperature for KNG, KNGS, and KNS glasses are shown in Fig. 2. It is well known that structural units of  $\text{NbO}_6$  octahedra,  $\text{GeO}_4$  and  $\text{SiO}_4$  tetrahedra in glasses give clear peaks in Raman scattering spectra, and, therefore, it would be worth summarizing features of Raman spectra of glasses containing  $\text{Nb}_2\text{O}_5$ ,  $\text{GeO}_2$  and  $\text{SiO}_2$  reported in many papers. According to Fukumi and Sakka [28,29], the Raman band in the  $800$ – $900\text{ cm}^{-1}$  region in glasses such as  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{SiO}_2$  is attributed to  $\text{NbO}_6$  octahedra with non-bridging oxygens and/or with much distortions and the broad bands in the  $600$ – $800\text{ cm}^{-1}$  region are attributed to less-distorted  $\text{NbO}_6$  octahedra with no non-bridging oxygens. For  $\text{GeO}_2$ - and  $\text{SiO}_2$ -based glasses, Raman

scattering spectra (i.e., peak position and intensity) relating to vibrations of  $\text{GeO}_4$  and  $\text{SiO}_4$  tetrahedra change significantly depending on the type and amount of network modifier cations. Basically, however, for  $\text{GeO}_2$ -based glasses, bands in the mid-frequency region of  $\sim 400$ – $\sim 500\text{ cm}^{-1}$  are assigned to stretching vibrations of  $\text{Ge}-\text{O}-\text{Ge}$  bonds (bridge) associated with four- and three-membered  $\text{GeO}_4$  rings, and bands in the high-frequency region of  $\sim 850\text{ cm}^{-1}$  are assigned to stretching vibrations of  $\text{Ge}-\text{O}^-$  bonds (terminal groups) [30–32]. In Raman scattering spectra for  $\text{SiO}_2$ -based glasses, analyzes similar to  $\text{GeO}_2$ -based glasses have been performed. That is, bands  $\sim 1100\text{ cm}^{-1}$  are assigned to stretching vibrations of  $\text{Si}-\text{O}^-$  bonds ( $\text{SiO}_4$  terminal groups with non-bridging oxygens) and bands in the  $400$ – $700\text{ cm}^{-1}$  region are assigned to bending vibrations of  $\text{Si}-\text{O}-\text{Si}$  linkages between  $\text{SiO}_4$  tetrahedra [33–35]. As the polarizability of  $\text{Nb}-\text{O}$  bonds is higher than that of  $\text{Si}-\text{O}$  bonds, generally the most intensive bands in Raman scattering spectra for  $\text{SiO}_2$ -based glasses with large amounts of  $\text{Nb}_2\text{O}_5$  are related to vibrational modes of  $\text{NbO}_6$  octahedra [15].

As can be seen in Fig. 2, peaks are observed at  $\sim 480\text{ cm}^{-1}$  for KNG and KNGS glasses containing  $\text{GeO}_2$ , but not in KNS glass. This peak would be,

therefore, assigned to stretching vibrations of Ge–O–Ge bonds between  $\text{GeO}_4$  tetrahedra. In KNG glass, a strong peak is observed at  $859\text{ cm}^{-1}$ , and this peak would indicate the presence of  $\text{GeO}_4$  tetrahedra with non-bridging oxygens. In KNGS and KNS glasses containing  $\text{SiO}_2$ , peaks are observed at  $\sim 1050\text{ cm}^{-1}$ , although their peak intensities are small. These peaks would be assigned to stretching vibrations of  $\text{Si–O}^-$  bonds. The Raman scattering spectra, therefore, indicate that  $\text{GeO}_4$  and  $\text{SiO}_4$  tetrahedra are clearly formed in  $\text{K}_2\text{O–Nb}_2\text{O}_5\text{–GeO}_2$  or  $\text{SiO}_2$  glasses with large amounts (50 mol%) of  $\text{K}_2\text{O}$  and  $\text{Nb}_2\text{O}_5$ . The peaks with similar intensities are observed at  $\sim 600\text{ cm}^{-1}$  for KNG, KNGS and KNS glasses, suggesting the presence of less-distorted  $\text{NbO}_6$  octahedra with no non-bridging oxygens. Furthermore, a strong shoulder for KNG glass and strong peaks for KNGS and KNS glasses are observed at  $900\text{ cm}^{-1}$ . Since no bands have been usually observed at  $\sim 900\text{ cm}^{-1}$  in silicate glasses, it would be reasonable to assign that these peaks at  $\sim 900\text{ cm}^{-1}$  are due to  $\text{NbO}_6$  octahedra, i.e.,  $\text{NbO}_6$  octahedra with non-bridging oxygens and/or with much distortions.

Kolobkova [36] reported the Raman scattering spectrum for  $25\text{K}_2\text{O–}25\text{Nb}_2\text{O}_5\text{–}50\text{GeO}_2$  glass and proposed the peak assignments similar to the present study. The Raman scattering spectrum for  $23\text{K}_2\text{O–}27\text{Nb}_2\text{O}_5\text{–}50\text{SiO}_2$  glass has been reported by Aronne et al. [15], and they concluded that the structure of the glass is formed by  $\text{SiO}_4$  tetrahedra and distorted  $\text{NbO}_6$  octahedra. In particular, they assigned that the Raman band at  $900\text{ cm}^{-1}$  is related to the vibration of highly distorted octahedral  $\text{NbO}_6$  or to octahedral  $\text{NbO}_6$  having at least a short terminal Nb–O bond pointing towards a modifier ion. Contrary, Alekseeva et al. [14] proposed that the Raman band at  $895\text{ cm}^{-1}$  observed for  $25\text{K}_2\text{O–}27.5\text{Nb}_2\text{O}_5\text{–}47.5\text{SiO}_2$  glass corresponds to the vibrations of  $\text{NbO}_4$  tetrahedra and other two bands at  $\sim 780$  and  $\sim 690\text{ cm}^{-1}$  are associated with the vibrations of the slightly distorted  $\text{NbO}_6$  octahedra with no non-bridging oxygen atoms. Although the Raman scattering spectra for  $25\text{K}_2\text{O–}25\text{Nb}_2\text{O}_5\text{–}50\text{SiO}_2$  (present study),  $23\text{K}_2\text{O–}27\text{Nb}_2\text{O}_5\text{–}50\text{SiO}_2$  [15], and  $25\text{K}_2\text{O–}27.5\text{Nb}_2\text{O}_5\text{–}47.5\text{SiO}_2$  [14] glasses have almost the same patterns, the interpretation of the band at  $\sim 900\text{ cm}^{-1}$  is still open to discussion. Considering the Raman scattering spectra for various niobium oxides with  $\text{NbO}_6$  and  $\text{NbO}_4$  coordinated units [37], however, it seems to be reasonable to assign that the peak at  $895\text{ cm}^{-1}$  observed for  $25\text{K}_2\text{O–}25\text{Nb}_2\text{O}_5\text{–}50\text{SiO}_2$  glass (Fig. 2) is related to distorted  $\text{NbO}_6$  octahedra, as proposed by Fukumi and Sakka [28,29] and Aronne et al. [15].

### 3.2. Formation of nanocrystals in $\text{KNb}(\text{Ge,Si})\text{O}_5$ glasses

KNG, KNGS, and KNS glasses were heat treated at their first crystallization peak temperatures for 1 h, and the results on the XRD analyzes for the powder (pulverized) samples are shown in Fig. 3. The broad peaks suggest the formation of nanoscaled crystals in these heat-treated

samples. The optical transmittance spectra for these crystallized glasses are shown in Fig. 4, indicating that the absorption edge is around 330 nm and the transmittance in the visible wavelength region is around 70%. As already clarified in previous studies [7–9], KNG glass shows a prominent bulk nanocrystallization, in which the crystalline phase is  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  ([38] and JCPDS 77-0963) and the size of nanocrystals is 10–30 nm. As an example, a transmittance electron microscope (TEM: JEM-3000F operating at 300 kV) image for the heat-treated ( $720^\circ\text{C}$ , 1 h) KNG sample is shown in Fig. 5. It is clearly confirmed the formation of nanocrystals with a diameter of  $\sim 20$  nm. In the previous papers [7,8], it has been proposed from the density measurements that the volume fraction of  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  nanocrystals formed in the bulk KNG crystallized ( $680^\circ\text{C}$ , 1 h) glass is about 60%. The results shown in Figs. 3–5 indicate that the nanocrystallization occurs not only in  $25\text{K}_2\text{O–}25\text{Nb}_2\text{O}_5\text{–}50\text{GeO}_2$  glass but also in  $25\text{K}_2\text{O–}25\text{Nb}_2\text{O}_5\text{–}25\text{GeO}_2\text{–}25\text{SiO}_2$  and  $25\text{K}_2\text{O–}25\text{Nb}_2\text{O}_5\text{–}50\text{SiO}_2$  glasses. Certainly, the formation of nanocrystals has been proposed in  $\text{K}_2\text{O–Nb}_2\text{O}_5\text{–SiO}_2$  glasses, although the structure and chemical composition of nanocrystals have not been clarified [10–15].

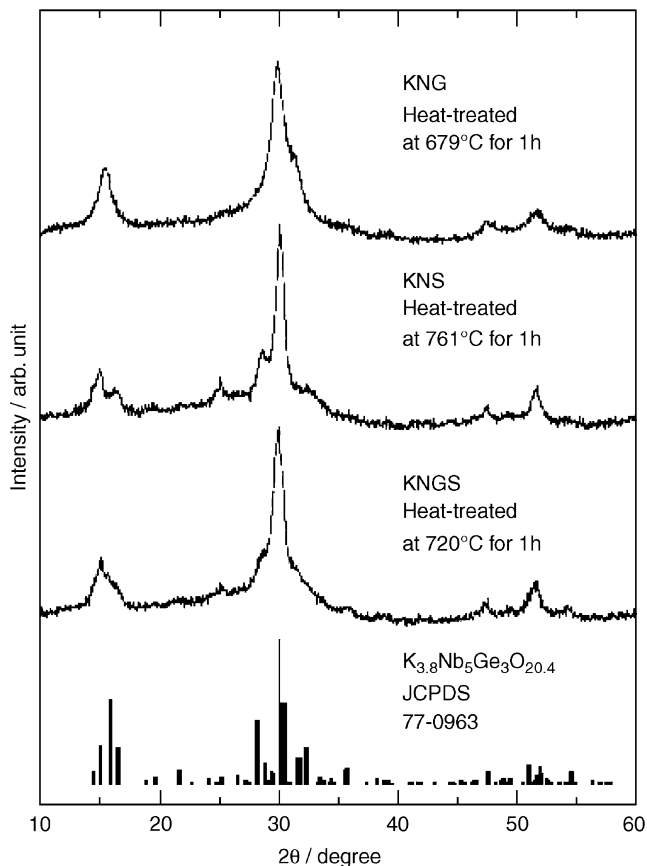


Fig. 3. Powder XRD patterns at room temperature for the crystallized glasses obtained by heat treatments at the crystallization peak temperatures for 1 h in the  $25\text{K}_2\text{O–}25\text{Nb}_2\text{O}_5\text{–}(50-x)\text{GeO}_2\text{–}x\text{SiO}_2$  system. KNG for  $x = 0$ , KNGS for  $x = 25$ , and KNS for  $x = 50$ .



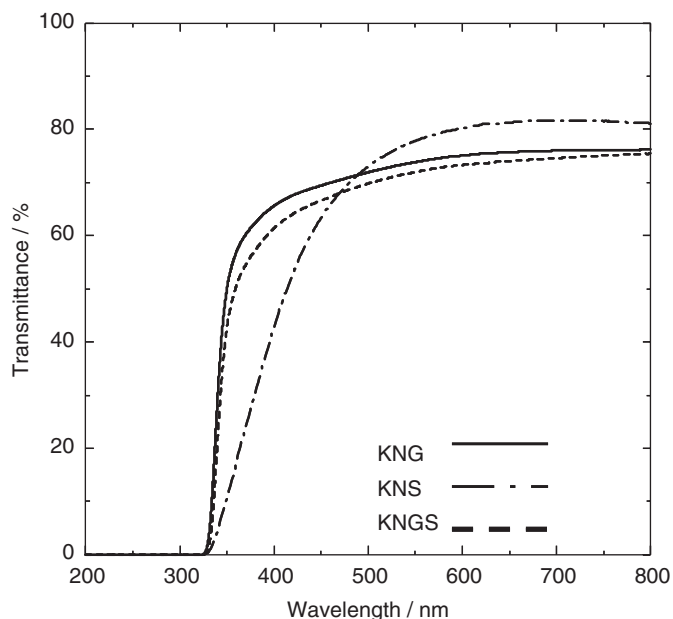


Fig. 4. Optical transmission spectra for the crystallized samples in the  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-(50-x)\text{GeO}_2-x\text{SiO}_2$  system. The heat-treatment temperatures are  $679^\circ\text{C}$  for KNG ( $x = 0$ ),  $720^\circ\text{C}$  for KNGS ( $x = 25$ ), and  $761^\circ\text{C}$  for KNS ( $x = 50$ ). The heat-treatment time was 1 h for all samples.

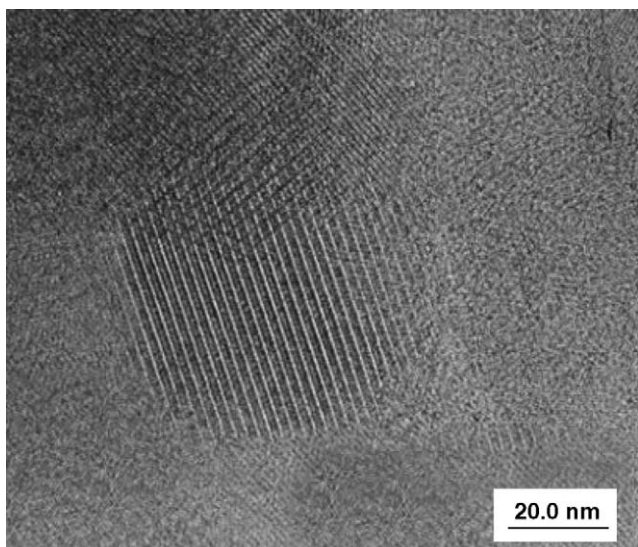


Fig. 5. TEM image for the crystallized ( $720^\circ\text{C}$ , 1 h) glass of  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-50\text{GeO}_2$ .

According to Belokoneva et al. [38], the  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  phase, which is prepared by cooling slowly (2 K/h) the melt of  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-50\text{GeO}_2$  in the temperature range of  $1100-700^\circ\text{C}$ , has an orthorhombic structure with the lattice constants of  $a = 0.7064$ ,  $b = 1.2177$ , and  $c = 2.2113$  nm. It is known that the structure of the  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  phase consists of three-membered tetrahedral rings formed by  $\text{GeO}_4$  tetrahedra,  $\text{NbO}_6$  octahedra linked to  $\text{Ge}_3\text{O}_9$  rings, and  $\text{NbO}_6$  octahedra connected to other  $\text{NbO}_6$  octahedra [38]. It should be pointed out that

$\text{GeO}_4$  tetrahedra are somewhat distorted and five different  $\text{NbO}_6$  octahedra are present [38].

As can be seen in Fig. 3, the XRD patterns for KNGS and KNS glasses are similar to KNG glass, suggesting that nanocrystals similar to  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$ , i.e.,  $\text{K}_{3.8}\text{Nb}_5(\text{Ge},\text{Si})_3\text{O}_{20.4}$  and  $\text{K}_{3.8}\text{Nb}_5\text{Si}_3\text{O}_{20.4}$ , might be formed due to the crystallization at  $T_p$ . The crystallization behavior of  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{SiO}_2$  glasses has been examined by some research groups [10–15], and it has been reported that unidentified (metastable) crystalline phases are formed due to the crystallization besides the stable phases of  $\text{K}_3\text{Nb}_3\text{Si}_2\text{O}_{13}$ ,  $\text{KNbSi}_2\text{O}_7$  and  $\text{KNbO}_3$ . The present study suggests that one of the metastable crystalline phases formed in  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{GeO}_2-\text{SiO}_2$  crystallized glasses is close to  $\text{K}_{3.8}\text{Nb}_5(\text{Ge},\text{Si})_3\text{O}_{20.4}$ .

The Raman scattering spectra for KNG, KNGS, and KNS crystallized glasses obtained by heat treatments at  $T_p$  for 1 h are shown in Fig. 6. It is seen that these spectra are considerably different from those (Fig. 2) for the precursor glasses. As a trend, the relative intensity of the bands observed at above  $700\text{ cm}^{-1}$  against the bands observed at below  $700\text{ cm}^{-1}$  is becoming small due to the nanocrystallization. Furthermore, it should be pointed out that nanocrystals formed in these heat-treated samples give

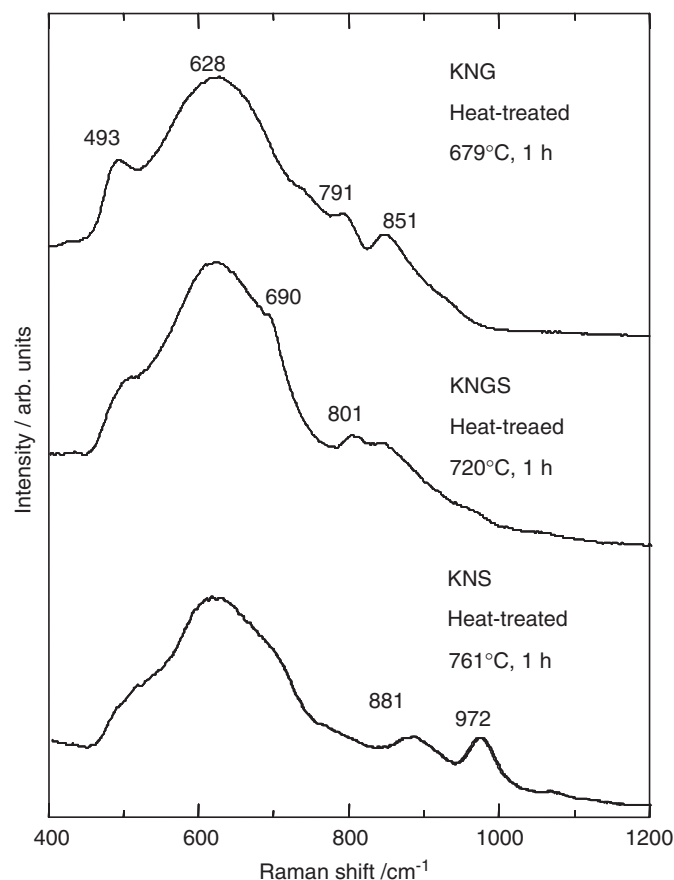


Fig. 6. Raman scattering spectra at room temperature for the crystallized glasses obtained by heat treatments at the crystallization peak temperatures for 1 h in the  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-(50-x)\text{GeO}_2-x\text{SiO}_2$  system. KNG for  $x = 0$ , KNGS for  $x = 25$ , and KNS for  $x = 50$ .

extremely broad Raman bands. In all samples, broad and strong bands are observed at around  $600\text{ cm}^{-1}$ . Considering the structure of  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  crystals, the main contribution to the bands at  $\sim 600\text{ cm}^{-1}$  would be due to the five different  $\text{NbO}_6$  octahedra. That is, it is considered that the amount of distorted  $\text{NbO}_6$  octahedra giving the Raman bands at around  $900\text{ cm}^{-1}$  observed in the precursor glasses (Fig. 2) decreases due to the nanocrystallization. The Raman spectra for the heat-treated sample of  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-50\text{SiO}_2$  glass (Fig. 6) are almost the same as those for  $23\text{K}_2\text{O}-27\text{Nb}_2\text{O}_5-50\text{SiO}_2$  [15], and  $25\text{K}_2\text{O}-27.5\text{Nb}_2\text{O}_5-47.5\text{SiO}_2$  [14] glasses. The SH intensity for these transparent crystallized (heat-treated at  $T_p$  for 1 h) glasses consisting of nanocrystals were examined, but no measurable SHGs were detected. It is, therefore, concluded that the  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  crystalline phase has an inversion symmetry in the structure, i.e., isotropic, and the substitution of  $\text{Ge}^{4+}$  for  $\text{Si}^{4+}$  do not break the inversion symmetry in the  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  phase.

The powder XRD patterns for the crystallized glasses obtained by heat treatments at temperatures ( $850$ ,  $900$ ,  $935^\circ\text{C}$ ) higher than  $T_p$  are shown in Fig. 7. In KNG sample, the crystalline phase of  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  is still present, and a relatively good transparency is still seen, as already reported in the previous papers [7,8]. But, in KNKS and KNS samples, the crystalline phase is not  $\text{K}_{3.8}\text{Nb}_5(\text{Ge,Si})_3\text{O}_{20.4}$ , and a transparency is disappeared.

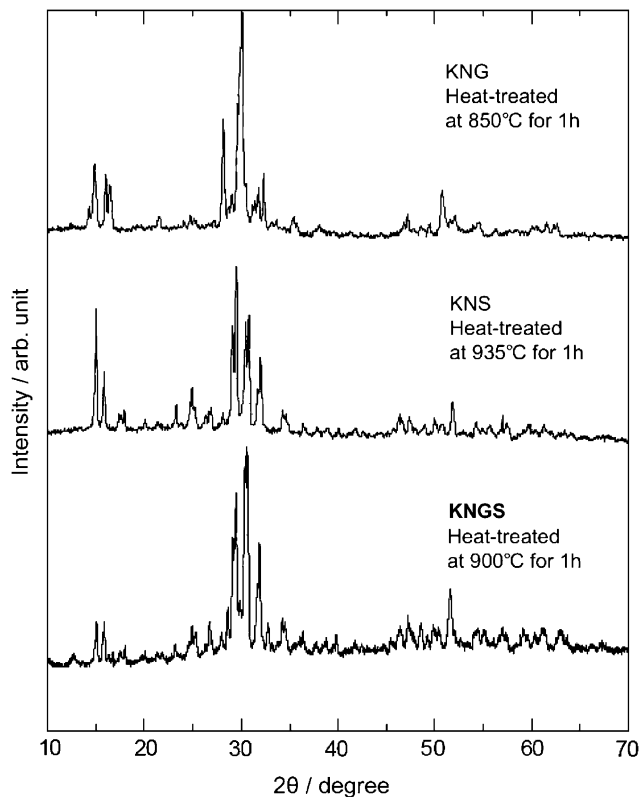


Fig. 7. Powder XRD patterns at room temperature for the crystallized glasses in the  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-(50-x)\text{GeO}_2-x\text{SiO}_2$  system. KNG for  $x = 0$ , KNKS for  $x = 25$ , and KNS for  $x = 50$ .

Unfortunately, the crystalline phases in these crystallized samples have not been identified at this moment, but the crystalline phases close to  $\text{K}_{3.8}\text{Nb}_5(\text{Ge,Si})_3\text{O}_{20.4}$ , i.e., the modified structure, are considered to be formed because of similar XRD patterns for these three samples as shown in Fig. 7. Since the transparency was disappeared for these crystallized KNKS and KNS glasses and the XRD peaks are relatively sharp, it is concluded that the growth of nanocrystals occurs in these KNKS and KNS samples. That is, it is considered that  $\text{SiO}_2$ -substituted  $\text{K}_{3.8}\text{Nb}_5(\text{Ge,Si})_3\text{O}_{20.4}$  nanocrystals might be unstable thermally in comparison with  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  nanocrystals.

From the above experiments, we clarified that all KNG, KNKS, and KNS glasses show nanocrystallization and transparent nanocrystallized glasses can be obtained through the crystallization at around the first crystallization peak temperatures. In particular, it was found that the nanocrystallization behavior of  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{GeO}_2-\text{SiO}_2$  glasses depends on the  $\text{GeO}_2/\text{SiO}_2$  ratio. These transparent nanocrystallized glasses do not show, however, any SHGs even for the substitution of  $\text{GeO}_2$  for  $\text{SiO}_2$ .

The dielectric constants (frequency: 1 kHz, temperature: room temperature, apparatus: TOYO corp. FCE-1 model),  $\epsilon$ , of the glasses and nanocrystallized glasses obtained by heat treatments at the first crystallization peak temperature for 1 h were measured, and the following values were obtained:  $\epsilon = 43, 41, \text{ and } 40$  for KNG, KNKS, and KNS glasses, respectively, and  $\epsilon = 58, 59, \text{ and } 49$  for KNG, KNKS, and KNS nanocrystallized glasses, respectively. These results clearly indicate that the dielectric constant of  $\text{K}_2\text{O}-\text{Nb}_2\text{O}_5-\text{GeO}_2-\text{SiO}_2$  glasses increases largely due to the nanocrystallization.

### 3.3. Chemical etching of nanocrystallized $\text{KNbGeO}_5$ glass fibers

KNG glass fibers with a diameter of  $\sim 100\ \mu\text{m}$  were prepared by inserting a silica glass rod into a melt and by pulling up the rod from the melt. The glass fibers obtained were heat treated at  $680^\circ\text{C}$  for 1 h. It was found that transparent nanocrystallized glass fibers are successfully fabricated as similar to the case of bulk plate glasses. Considering the volume fraction of  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  nanocrystals formed in the bulk (not fibers) KNG crystallized ( $680^\circ\text{C}$ , 1 h) glass [7,8], i.e., 60%, it would be expected that the crystallized glass fibers also have a similar volume fraction (60%) of  $\text{K}_{3.8}\text{Nb}_5\text{Ge}_3\text{O}_{20.4}$  nanocrystals. The base glass fibers and nanocrystallized glass fibers were etched using a meniscus chemical etching method [18,20,21,24], in which the etching solutions of 10 wt% -HF/hexane and 10 M -NaOH/hexane were used. In this etching technique, it is expected that the evolution of a meniscus at the interface between the fiber and the solution leads to the tip formation, and hexane is used as a protecting overlayer against HF or NaOH etching solution. It is well recognized that network structures such as Si-O-Si bonds in oxide glasses are attacked and broken by acid solutions such as

$F^-$  ions in HF or alkali solutions such as  $OH^-$  ions in NaOH. There has been no report on chemical etching behaviors in  $GeO_2$ -based glasses and crystallized glasses, and thus we tried to use both aqueous HF and NaOH solutions in this study.

The polarization optical microphotographs for the KNG glass fiber etched chemically using the 10 wt%-HF/hexane system are shown in Fig. 8. The temperature of the etching solution was fixed to room temperature, and the etching time was 2 h. It is seen that the taper length is about 270  $\mu m$  and the tip angle is about 23°. Although the tip (aperture size) is not a nanoscaled size, the data shown in Fig. 8 demonstrate that the 10 wt%-HF/hexane system works as an etching solution for KNG glass fibers. The polarization optical microphotographs for the transparent KNG nanocrystallized glass fiber etched chemically using the 10 wt%-HF/hexane system are shown in Fig. 9. It is clear that the etched shape is different from that for the precursor glass fiber (Fig. 8). That is, the taper length is about 450  $\mu m$  and the tip angle is about 12°. These results indicate that the meniscus state and etching rate for the nanocrystallized KNG glass fibers in the 10 wt%-HF/hexane system is different from those for the precursor KNG glass fibers. In other words, it is considered that nanocrystals themselves and grain boundaries among

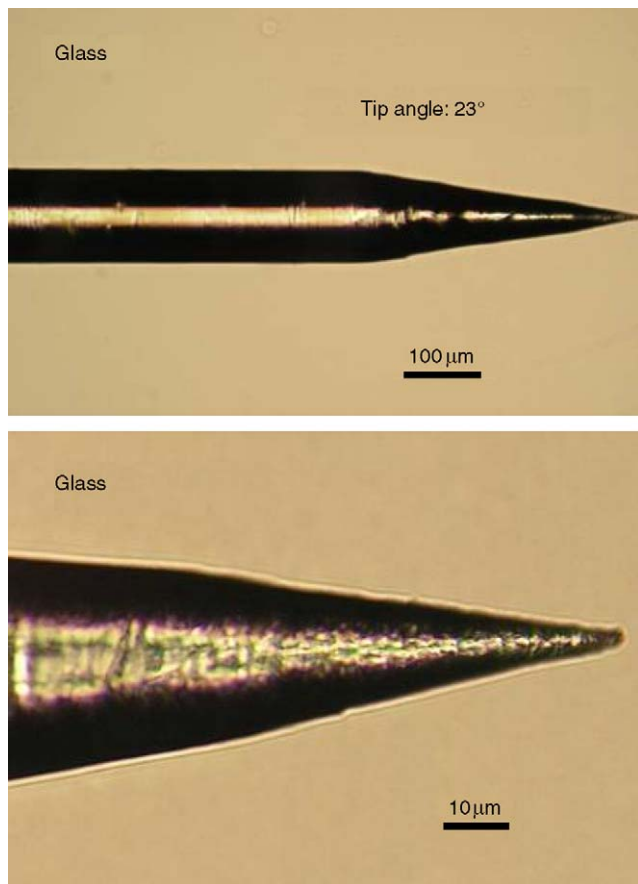


Fig. 8. Polarization optical microphotographs for the glass fiber etched (room temperature, 2 h) chemically using the 10 wt%-HF/hexane system. The glass composition is  $25K_2O-25Nb_2O_5-50GeO_2$ .

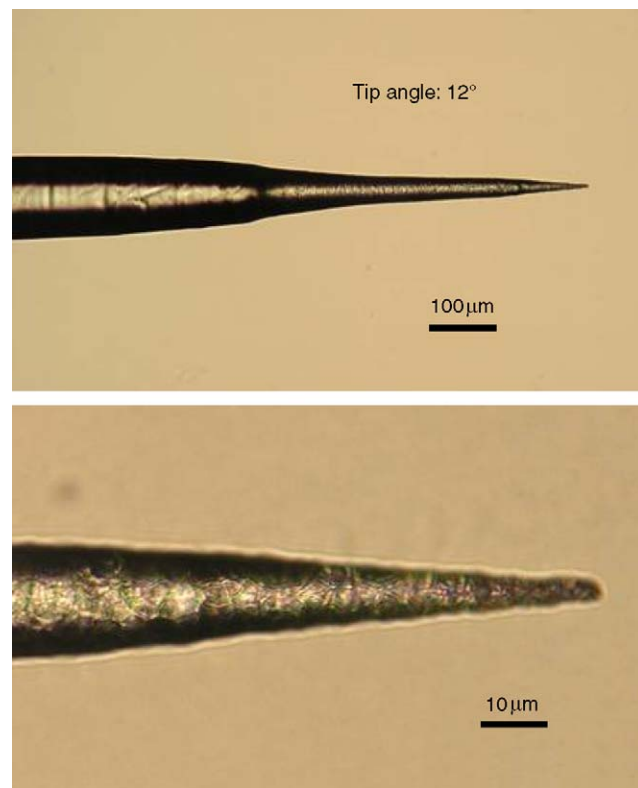


Fig. 9. Polarization optical microphotographs for the nanocrystallized glass fiber etched (room temperature, 2 h) chemically using the 10 wt%-HF/hexane system. The glass composition is  $25K_2O-25Nb_2O_5-50GeO_2$ .

nanocrystals in nanocrystallized glass fibers affects largely the etching behavior.

The polarization optical microphotographs for the transparent KNG nanocrystallized glass fiber etched chemically (at room temperature for 2 weeks) using the 10 M-NaOH/hexane system are shown in Fig. 10. It was found that the etching rate is very low, and a long etching time, i.e., 2 weeks, was needed to get sharpened fibers. As shown in Fig. 10, the taper length is about 2 mm, the tip angle is about 3°, and the shape of fiber tip is not sharp. These results indicate that the 10 M-NaOH/hexane system is not so effective for the etching of KNG nanocrystallized glass fibers compared with the 10 wt%-HF/hexane system.

It is known that the optical resolution in SNOM is of the order of the size of the probe aperture [39,40], and thus it is important to fabricate optical fiber probes with a nanoscaled aperture. It is desired to examine the etching behavior of transparent nanocrystallized glass fibers in various etching solutions and to find optimal etching solutions and conditions for the fabrication of fiber probes with a nanoscaled aperture. The transparent  $KNb(Ge,Si)O_5$  nanocrystallized glasses have the refractive indices of  $n = 1.75-1.81$  and the dielectric constants of  $\epsilon = 50-59$ , being much higher than those ( $n = 1.45$ ,  $\epsilon = 4$ ) for  $SiO_2$  glass. This would be a strong motivation for the study of nanocrystallized glass fiber probes in the  $K_2O-Nb_2O_5-GeO_2-SiO_2$  glass system.



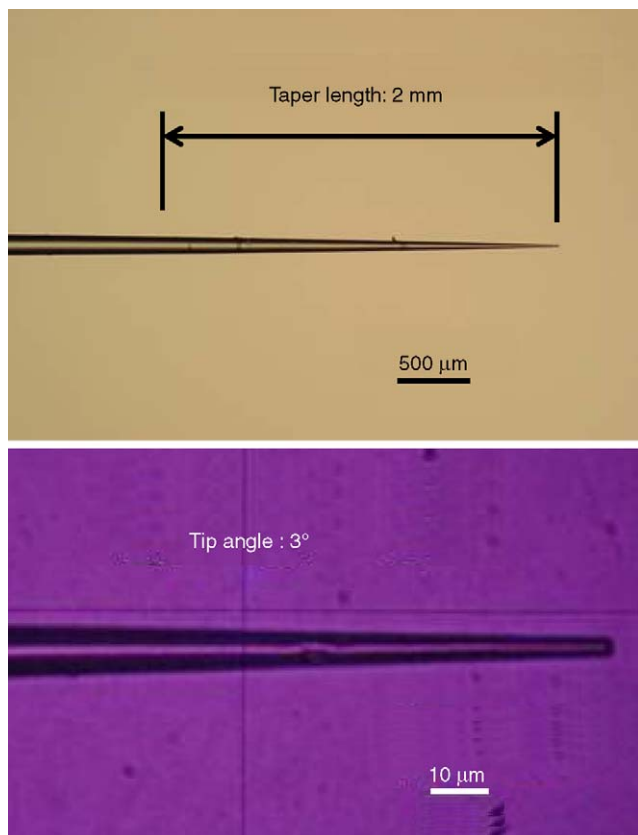


Fig. 10. Polarization optical microphotographs for the nanocrystallized glass fiber etched (room temperature, 2 weeks) chemically using the 10 M-NaOH/hexane system. The glass composition is  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-50\text{GeO}_2$ .

The mechanical properties such as Young's modulus and Vickers hardness of transparent KNG nanocrystallized glass fibers fabricated in this study have not been clarified at this moment. In the previous papers [7,8], the author group clarified that the elastic and mechanical properties of bulk KNG glass are largely improved due to the nanocrystallization. For instance, KNG glass has the values of Young's modulus  $E = 64$  GPa and Debye temperature  $\theta_D = 500$  K, but the transparent nanocrystallized (heat-treated at  $630^\circ\text{C}$ , 1 h) glass shows  $E = 81$  GPa and  $\theta_D = 564$  K. It is expected that transparent KNG nanocrystallized glass fibers would have elastic and mechanical properties similar to bulk transparent KNG nanocrystallized bulk glasses.

#### 4. Conclusion

The crystallization behavior of  $25\text{K}_2\text{O}-25\text{Nb}_2\text{O}_5-(50-x)\text{GeO}_2-x\text{SiO}_2$  glasses with  $x = 0, 25$ , and  $50$  (i.e.,  $\text{KNb}(\text{Ge},\text{Si})\text{O}_5$  glasses) was examined to fabricate transparent nanocrystallized glasses showing SHGs. In all compositions, transparent nanocrystallized glasses were obtained through the crystallization at first crystallization peak temperatures, although they did not show measurable SHGs. The glass fibers with a diameter of about  $100\ \mu\text{m}$

were fabricated by just pulling up the melt using a silica rod, and transparent nanocrystallized glass fibers were produced by heat treatments of the precursor glass fibers. The fibers with sharpened tips (e.g., the taper length is about  $450\ \mu\text{m}$  and the tip angle is about  $12^\circ$ ) were obtained using a meniscus chemical etching method, in which the etching solutions of 10 wt%-HF/hexane and 10 M-NaOH/hexane were used.

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